This article was downloaded by:

On: 27 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK

Organ Publica http://w

Associate Editor

AND PROCEDURES INTERNATIONAL

> Executive Editor J.-P. ANSILME sociate Executive Edito M. J. HEARN

Taylor & Francis

Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t902189982

SYNTHESIS OF 4,4'-DIHYDROXY-3,3',5,5'-TETRA-t-BUTYLBENZOPHENONE

Mary K. Connera; E. Legoffa

^a Department of Chemistry, Michigan State University, East Lansing, Michigan

To cite this Article Conner, Mary K. and Legoff, E.(1970) 'SYNTHESIS OF 4,4'-DIHYDROXY-3,3',5,5'-TETRA-t-BUTYLBENZOPHENONE', Organic Preparations and Procedures International, 2: 1, 69-70

To link to this Article: DOI: 10.1080/00304947009458422 URL: http://dx.doi.org/10.1080/00304947009458422

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

SYNTHESIS OF 4,4'-DIHYDROXY-3,3',5,5'-TETRA-t-BUTYLBENZOPHENONE

Mary K. Conner and E. LeGoff¹
Department of Chemistry, Michigan State University
East Lansing, Michigan 48823

Several attempts on our part to prepare the substituted benzophenone \mathfrak{F} using standard Friedel-Crafts procedures (2,6-di-t-butylphenol with carbon tetrachloride or phosgene and aluminum chloride or stannic chloride) have been unsuccessful. The starting material and products readily lost the \underline{t} -butyl moeity affording complex mixtures.

We find, however, that a simple, oxidative procedure will give good yields of \mathfrak{F} . Acetic acid-sulfuric acid solutions of 2,6,3',5'-tetra- \underline{t} -butyl-4'-hydroxyphenyl-4-methylene-2,5-cyclohexadiene-1-one (\mathfrak{F}) or 4,4'-dihydroxy-3,3',5,5'-tetra- \underline{t} -butyldiphenylethoxymethane (\mathfrak{F}) undergo air oxidation giving 4,4'-dihydroxy-3,3',5,5'-tetra- \underline{t} -butylbenzophenone (\mathfrak{F}) in about 80% yield.

HO
$$\downarrow$$
 CH= \downarrow O
 \uparrow HOAc - H₂SO₄ HO \downarrow CO \downarrow OH
 \downarrow OC₂H₅ OH
 \uparrow OH

EXPERIMENTAL

4,4'-Dihydroxy-3,3',5,5'-tetra- \underline{t} -butyldiphenylethoxymethane (2). - A solution of 3 g of 4,4'-dihydroxy-3,3',5,5'-tetra- \underline{t} -butyldiphenylbromomethane² in 20 ml of absolute etahnol and 1 ml of hydrochloric acid was refluxed for 2 hrs. Upon cooling, 1.5 g (yield 54%) of yellow product was obtained. Two recrystallizations from ethanol and one from pentane gave a colorless product: mp 162-163°; ir (CHCl₃) 3640 (hindered hydroxyl) and 1075 cm⁻¹ (ether); nmr (CDCl₃) δ 7.11 (s, 4H), 5.21 (s, 1H), 5.07 (s, 2H), 3.51 (q, 2H, J = 7 Hz), 1.41 (s, 36H) and 1.25 ppm (t, 3H, J = 7 Hz).

Anal. Calcd. for C₃₁H₄₈O₃: C, 79.49; H, 10.25;

Found: C, 79.38; H, 10.23.

4,4'-Dihydroxy-3,3',5,5'-tetra- \underline{t} -butylbenzophenone (\mathfrak{J}). - a) From ethoxy compound, \mathfrak{L} : A solution of 0.5 g of ethoxy compound \mathfrak{L} in 5 ml of acetic acid containing 3 drops of concentrated sulfuric acid was stirred in an opened flask for 24 hrs.³ The solid which separated was filterd. Water was slowly added to the filtrate to give additional product. The combined solids were dried and crystallized from carbon tetrachloride to give 0.38 g (81%) of benzophenone \mathfrak{J} : mp 221-222°; 2,4-DNP mp 180-182; ir (CHCl₃) 3620 (hindered hydroxyl) and 1635 cm⁻¹ (C=0); nmr (CDCl₃) δ 7.75 (s, 2H), 5.70 (s, 1H) and 1.48 ppm (s, 18H).

Anal. Calcd. for C₂₉H₄₂O₃: C, 79.50; H, 9.59; Found: C. 79.45; H. 9.65.

b) From cyclohexadieneone 1^2 : Following the same procedure as in a 2 g of cyclohexadieneone 1 in 25 ml of acetic acid containing 10 drops of sulfuric acid gave 1.6 g (78% yield) of benzophenone 3.

REFERENCES

- 1. Author to whom correspondence should be addressed.
- 2. M. S. Kharasch, B. S. Joshi, J. Org. Chem., 22, 1435 (1957).
- 3. When air (oxygen) was excluded from the reaction the yield was very poor.

<u>Acknowledgement</u>: Financial support from the National Institutes of Health (Grant AI-07869) is gratefully acknowledged.

(Received October 14, 1969)